## **Amendments to the Claims**

This listing of claims will replace all prior versions and listings of claims in the application.

## **Listing of Claims**

1. (Original) Substantially pure deslorated ine having an HPLC purity greater than 99.5%, and having an absorbance less than 0.15 Au at 420 nm for a 5%w/v solution in methanol, which does not show a peak for an impurity at a relative retention time in the range from about 0.85 to about 0.99 (relative to deslorated appearing at a retention time of 25±5 minutes) which is greater than the discard limit set at less than 0.025% of total area, when tested according to an HPLC method performed using a Hypersil BDS C<sub>8</sub> column (15 cm x 4.6 mm, 5 μm particle size) with the following parameters:

Mobile phase: Buffer solution having a pH of about 3, methanol and acetonitrile in a volume ratio of 8:1:1.

Injection volume : 20µl

Flow rate : 1.5 ml/minute

Run time : 75 minutes

Discard limit : Set at less than 0.025% of total area

- 2. (Original) Substantially pure desloratedine as claimed in claim 1, wherein (a) total impurities are not more than 0.5%; and (b) no individual impurity is greater than 0.1%.
- 3. (Original) Substantially pure desloratedine as claimed in claim 2, wherein the total impurities are less than 0.3%.
- 4. (Previously presented) Substantially pure desloratedine of claim 1, 2 or 3 prepared by a process comprising acidic hydrolysis of a compound of formula 3, where R is selected from COR<sub>1</sub>, COOR<sub>1</sub>, wherein R<sub>1</sub> is selected from branched or linear alkyl containing 1

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to 6 carbon atoms, cycloalkyl, alkenyl, alkynyl, aryl, aralkyl and their substituted analogs; the acidic hydrolysis comprising

heating with a strong organic acid or a mineral acid for about 1 hour to about 24 hours,

adjusting pH of the hydrolysed reaction mixture to about 3 to 5, optionally treating with an adsorbent, adjusting the pH of the reaction mixture to greater than about 9, and isolating desloratedine

Formula 3.

- 5. (Previously presented) The substantially pure desloratedine of claim 4 prepared by a process comprising heating to about 20° to 150°C in the presence of a strong acid.
- 6. (Previously presented) The substantially pure desloratedine of claim 4 prepared by a process further comprising recrystallizing desloratedine from a solvent system comprising a mixture of an alcohol and a hydrocarbon solvent.
- 7. (Previously presented) The substantially pure desloratedine of claim 6 wherein the alcohol is methanol and the hydrocarbon solvent is cyclohexane.
- 8. (Previously presented) A process for preparation of substantially pure desloratedine comprising acidic hydrolysis of a compound of formula 3, where R is selected

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from COR<sub>1</sub>, COOR<sub>1</sub>, wherein R<sub>1</sub> is selected from branched or linear alkyl containing 1 to 6 carbon atoms, cycloalkyl, alkenyl, alkynyl, aryl, aralkyl and their substituted analogs; comprising heating the compound of formula 3 in the presence of with-a strong organic acid or a mineral acid for about 1 hour to about 24 hours,

adjusting the pH of the hydrolysed reaction mixture to about 3 to 5, optionally treating the pH adjusted reaction mixture with an adsorbent, adjusting the pH of the reaction mixture to greater than about 9, and isolating desloratadine

Formula 3.

- 9. (Previously presented) The process as claimed in claim 8 wherein R is  $COOR_1$  and  $R_1$  is ethyl and the acid is methanesulfonic acid.
- 10. (Previously presented) The process as claimed in claim 8 wherein R is  $COOR_1$  and  $R_1$  is ethyl and the acid is sulphuric acid.
- 11. (Previously presented) The process as claimed in claim 8, comprising heating to about 20° to 150°C in the presence of a strong acid.
- 12. (Previously presented) The process as claimed in claim 11, comprising heating to about 60°C to 110°C in the presence of a strong acid.

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- 13. (Previously presented) The process as claimed in claim 9, comprising heating with methanesulfonic acid for 5 to 15 hours at about 90°C to 120°C.
- 14. (Previously presented) The process as claimed in claim 10, comprising heating with sulphuric acid for 1 to 5 hours at about 90°C to 120°C.
- 15. (Previously presented) The process as claimed in claim 8, wherein adsorbent is selected from charcoal, neutral or alkaline alumina, silica and fuller's earth.
- 16. (Previously presented) The process as claimed in claim 8, comprising adjusting the pH of the reaction mixture to about 4 to 5, treating with charcoal, adjusting the pH of the reaction mixture to greater than about 9 and isolating desloratedine.
- 17. (Previously presented) The process as claimed in claim 8, further comprising recrystallizing desloratedine from a solvent system comprising of two or more solvents selected from water, alcohols, linear hydrocarbons, branched hydrocarbons, cyclic hydrocarbons, aromatic hydrocarbons, ethers, ketones, nitriles, esters, and their halo or substituted analogs-and the like.
- 18. (Previously presented) The process as claimed in claim 8, further comprising recrystallizing desloratedine from a solvent system comprising a mixture of an alcohol and a hydrocarbon solvent.
- 19. (Previously presented) The process as claimed in claim 18 wherein alcohol is methanol and hydrocarbon solvent is cyclohexane.
- 20. (Previously presented) The process as claimed in claim 19, wherein the ratio of methanol:cyclohexane is 1:14 v/v.

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- 21. (Previously presented) The process as claimed in claim 8 for preparation of substantially pure desloratedine as described in claim 1, 2 or 3.
  - 22. (Canceled)